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**balance the atoms
and charge in a
chemical reaction,
you can examine
the energy of the
reaction as well.**

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Structure: And

**Electrons are
found in regions
around the
nucleus of an
atom.**

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**Chemistry studies
changes in matter.**

**A chemical
reaction is a**

**process in which
one set of
chemical**

compounds are

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**transformed into
another. Reaction
occur when there
is an interaction
between the
compounds in
which some initial
bonds are broken
and some new
bonds are formed.**

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effort to
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Orbitals. A lot of

chemistry is

explained by the

sharing and

trading of

electrons between

atoms.

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Understanding how electrons are arranged in an atom is a building block of Chem I. Electrons in an atom are contained in specific energy levels (1, 2, 3, and so on) that are different distances from the nucleus.

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chemistry A
physical quantity
is expressed in
terms of pure
number and unit
Physical quantity =
(a pure number) ×**

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when we say 5 kg.
It means 5 times of
1kg 'A unit is
defined as the
standard of
reference chosen
in order to
measure a definite
physical quantity'.**

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the fundamental
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physics,
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biology. Energy is
the ability to bring
about change or to
do work. It exists
in many forms,
such as heat, light,**

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**chemical energy,
and electrical
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energy can be
captured in
chemical bonds,
such as in the
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physical quantity
is expressed in
terms of pure
number and unit
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(a pure number) ×
unit For example
when we say 5 kg.
It means 5 times of
1kg 'A unit is
defined as the
standard of**

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in order to
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measure a definite
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physical quantity”.

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concepts in
physics,**

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biology. Energy is
the ability to bring
about change or to
do work. It exists
in many forms,
such as heat, light,
chemical energy,
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energy. Light
energy can be
captured in
chemical bonds,**

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not, you must**

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concepts of**

Chemistry. I'm

**sure after learning
these, you might**

**become a fan of
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