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There are various forms of potential energy such as gravitational, elastic, nuclear, electric and chemical potential. Here we will focus on the latter. Bonding with Chemical Potential Energy. Chemical potential energy is that which is stored within the bonds of atoms and molecules of a substance. The bonds have stored chemical potential energy which is released when they are broken and the substance undergoes a chemical reaction.

What is Chemical Potential Energy? A Study Guide ...

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File Type PDF Study Guide Energy And Chemical Change Answers Chemical potential energy depends on the strength of the bonds; strong bonds have low energy and weak ones have high energy. Large amounts of heat and/or light energy are released during the formation of strong bonds. Conversely, more energy is required to break strong bonds than weaker

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A spark called the energy of activation is needed. The activation energy needed to spark an exergonic (energy-yielding) reaction or endergonic (energy-requiring) reaction can be heat energy or chemical energy. Reactions that require activation energy can also proceed in the presence of biological catalysts. Catalysts are substances that speed up chemical reactions but remain unchanged during the reactions.

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Atoms are held together by a certain amount of energy called bond energy. Chemical processes are labeled as exothermic or endothermic based on whether they give off or absorb energy, respectively. 7.5: The Energy of

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