

Periodic Trends Review Notes Sheet Answers

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Uses of the Periodic Table Useful in predicting: chemical behavior of the elements trends properties of the elements Atomic Structure Review: Atoms are made of protons, electrons, and neutrons. Elements are atoms of only one type. Elements are identified by the atomic number (# of protons in nucleus).

The Periodic Table of The Elements

Periodic Trends Multiple Choice Review PSI Chemistry Name _____ Atomic Size 1) Elements Z and X are compared. Element Z is larger than Element X. Based on this you could say: A) Element Z is further to the left side of the periodic table B) Element X is closer to the top of the periodic table C) Element Z and X are probably in the same group ...

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NOTES: 6.3 –Periodic Trends. PERIODIC TABLE: RECALL... Periodicity:regular variations (or patterns) of properties with increasing atomic weight; both chemical and physical properties vary in a periodic (repeating pattern). Group:vertical column of elements (“ family ”) Period:horizontal row of elements.

CHAPTER 6 NOTES: The Periodic Table

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Periodic Table Activity: Complete the table on page 21 with the information found on pages 18-20. When complete color each group in a different color in the periodic table. The Periodic Table Notes: Historical development of the periodic table: Highlights • Mendeleev (1869): Put the elements into columns according to their properties.

Unit 3 Notes: Periodic Table Notes

The noble gases (Group 18) are located in the far right of the periodic table and were previously referred to as the "inert gases" due to the fact that their filled valence shells (octets) make them extremely nonreactive. The noble gases were characterized relatively late compared to other element groups. They are not used when referring to trends.

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