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Hydrogen is in group 1 and period 1. Aluminum is in the group 13 but group 3 for the main group elements, the period is 3. Silver is Ag which is in group 11 and period 5. Identify which elements are considered as the main group or transition metals.

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Organization of the Periodic Table. The periodic table lists elements by atomic number, which is the number of protons in every atom of that element. Atoms of an atomic number may have varying numbers of neutrons (isotopes) and electrons (ions), yet remain the same chemical element. Elements in the periodic table are arranged in periods (rows) and groups (columns).

## Introduction to the Periodic Table of Elements

A. There are 2 electrons in the first energy level, 8 protons in the second energy level and 8 neutrons in the third energy level. B. Refers to the arrangement of electrons in the electron cloud. C. There are two electrons in the first energy level, a maximum of eight electrons in the second and third energy levels. D.

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