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and study of the material**

**world. It is generally accepted that there are three states of matter, solid, liquid and gaseous, and the chemicals that make up the materials of the world involve the**

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**chemical elements or  
molecules. ELEMENTS,  
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are some important  
fundamental concepts that  
we need to understand  
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**concepts are very useful for dealing with the organic reactions in later chapters. Nucleophiles: A nucleophile is a reagent that is either negatively charged or carry lone pair of electrons.**

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**These reagents ...**

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**chemical bond that involves  
the sharing of electron  
pairs between atoms.**

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**e. 1. Branches of chemistry.**  
**Organic chemistry :** This branch deals with study of compounds of carbon and its compounds with any number of other elements, including hydrogen (most

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**compounds contain at least one carbon-hydrogen bond), nitrogen, oxygen, halogens, phosphorus, silicon, and sulfur. Except carbonates, bicarbonates, cyanides, isocyanides,**



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**carbides and oxides**

**Inorganic chemistry: This branch deals with study of known elements and their compounds except organic compounds.**

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Concept The substituents in  
the aromatic ring define its  
reactivity towards**

**electrophilic aromatic substitution. As we mentioned, the aromatic ring acts as a nucleophile. Therefore, it will be much more reactive if it has substituents that donate**

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**electrons to the ring.**

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Chemistry is called the  
science of atoms and**

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**molecule Branches of  
Chemistry Organic  
Chemistry -This branch  
deals with study of carbon  
compounds especially  
hydrocarbons and their  
derivatives. Inorganic**

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**Chemistry-This branch deals with the study of compounds of all other elements except carbon.**

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and Organic Chemistry.  
While inorganic chemistry  
includes all elements,  
organic chemistry studies  
compounds of Carbon. Why**

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**is carbon so important that  
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provided us with new  
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ELECTRONEGATIVITY A  
covalent bond, where the  
electrons are shared equally  
is called a nonpolar bond**

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**(eg H-H) and an unequal sharing of the pair of bonding electrons results in a polar bond. The unequal sharing of electrons is due to the ability of an atom to attract electrons towards**

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**itself which is known as  
Electronegativity.**

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Tetravalent carbon atom is  
the building block of  
structural organic  
chemistry. The four**

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**hydrogen atoms, with four  
carbon atoms, form a  
structure known as a  
tetrahedron.**

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chemistry is the study of  
compounds containing  
carbon with the exception  
of simple compounds  
e.g. carbonates ( $\text{CO}_3^{2-}$ ),**

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**carbon dioxide (CO<sub>2</sub>) and  
carbon monoxide (CO).**

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**The concepts treated in the  
seven programmes are**

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**fundamental to any  
introductory course in  
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include certain intellectual  
techniques such as "curly  
arrows" which students  
often find difficult, but**

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**which need to be fully mastered. The programmes are built up from frames. Each frame tells something about the concept and ...**

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**carefully. These basic concepts are very useful for dealing with the organic reactions in later chapters. Nucleophiles: A nucleophile is a reagent that is either negatively charged or carry**



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**lone pair of electrons.**

**These reagents ...**

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**Chemistry ...**

**e. 1. Branches of chemistry.**  
**Organic chemistry : This**  
**branch deals with study of**  
**compounds of carbon and**  
**its compounds with any**  
**number of other elements,**

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**including hydrogen (most  
compounds contain at least  
one carbon-hydrogen  
bond), nitrogen, oxygen,  
halogens, phosphorus,  
silicon, and sulfur. Except  
carbonates, bicarbonates,**

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**cyanides, isocyanides,  
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**reactivity towards  
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mentioned, the aromatic  
ring acts as a nucleophile.  
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**substituents that donate  
electrons to the ring.**

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# **CONCEPTS OF CHEMISTRY**

## **Chemistry**

**Yes. They are- Inorganic  
and Organic Chemistry.**

**While inorganic chemistry  
includes all elements,  
organic chemistry studies**

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**compounds of Carbon. Why  
is carbon so important that  
it is studied under a  
different branch?**

**Organic Chemistry: Some  
Basic Principles and**

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**Techniques ...**

**Basic Concepts from  
Organic Chemistry (PDF  
103p) This note covers the  
following topics: Aliphatic  
Compounds, Hydrocarbons,  
Alcohols, Aldehydes And**

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**Ketones, Acids, Esters,  
Ethers, Alkyl Halides And  
Other Halogenated  
Aliphatic Compounds,  
Simple Compounds  
Containing Nitrogen, Cyclic  
Aliphatic Compounds,**

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**Mercaptans Or  
Thioalcohols, Aromatic  
Compounds, Hydrocarbons  
and Phenols.**

**Basic Concepts from  
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**103p ...**

**Chemistry is often called the central science, because a basic knowledge of chemistry is essential for students in biology, physics, geology and many other**

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**subjects. Chemical research and development has provided us with new substances with specific properties. These substances have improved the quality of our lives.**

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**General Inorganic  
Chemistry - Matlidents  
Gymnasium  
ELECTRONEGATIVITY A  
covalent bond, where the  
electrons are shared equally**

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**is called a nonpolar bond (eg H-H) and an unequal sharing of the pair of bonding electrons results in a polar bond. The unequal sharing of electrons is due to the ability of an atom to**

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**attract electrons towards  
itself which is known as  
Electronegativity.**

**Basic Concepts of Organic  
Chemistry | Chemistry  
Notes for ...**

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**Fundamental Concepts in  
Organic Reaction  
Mechanism The Shapes of  
Carbon Compounds -  
Tetravalent carbon atom is  
the building block of  
structural organic**

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**chemistry. The four hydrogen atoms, with four carbon atoms, form a structure known as a tetrahedron.**

**Organic Chemistry Class 11**

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**- Basic Principles &  
Techniques**

**Organic Chemistry Organic  
chemistry is the study of  
compounds containing  
carbon with the exception  
of simple compounds**



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**e.g. carbonates ( $\text{CO}_3^{2-}$ ),  
carbon dioxide ( $\text{CO}_2$ ) and  
carbon monoxide ( $\text{CO}$ ).**

**An Introduction to Organic  
Chemistry  
The concepts treated in the**

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**seven programmes are  
fundamental to any  
introductory course in  
Organic Chemistry and  
include certain intellectual  
techniques such as "curly  
arrows" which students**

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**often find difficult, but  
which need to be fully  
mastered. The programmes  
are built up from frames.  
Each frame tells something  
about the concept and ...**