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**generally do a**

**good job of**

**determining**

**vapor-phase**

**properties. But**

**accurate**

**determination of**

**properties for the**

**liquid phase is**

**much more**

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difficult,

especially when

the liquid

contains

dissimilar

molecules, polar

molecules such

as alcohols and

glycols, or acid

gasses such as

H<sub>2</sub>S and CO<sub>2</sub>.

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like alcohol-  
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design many  
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**mixtures, it is**

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**conductivities,**

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**much more**

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difficult,

especially when

the liquid

contains

dissimilar

molecules, polar

molecules such

as alcohols and

glycols, or acid

gasses such as

H<sub>2</sub>S and CO<sub>2</sub>.

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(VLE) data of

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**conductivity of**

**multicomponent**

**mixtures, it is**

**necessary to use**

**a mixing rule that**

**utilizes the**

**conductivities,**

**either**

**experimentally**

**obtained or**

**calculated, of**

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binary

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containing one

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mixing rules can

be written in

terms of molar

fractions or

solute fractions

with equivalent

ionic strength

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