

Chapter 8 Review Chemical Equations And Reactions

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-One element in the reactant of a chemical compound replaces one element in the product of a Chemical Reaction -General Eqn= $A + Bx = AX + B - Ex = Fe + Cu(NO_3)_2 \rightarrow Fe(NO_3)_2 + Cu$ -Note cations will replace cations while anions will replace anions

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CHAPTER 8 REVIEW. Chemical Equations and Reactions. SECTION 2. SHORT ANSWER Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right. (4 points) _____ synthesis _____ decomposition _____ single-displacement _____ double-displacement. 2. _____ In the reaction described by the equation $2\text{Al}(s)$

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This results in a correct chemical equation, or balanced formula equation, for the burning of methane in oxygen. $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{g})$ This reaction is further illustrated in Figure 8-3. Additional Symbols Used in Chemical Equations Table 8-2 on page 246 summarizes the symbols commonly used in chemi-CHEMICAL EQUATIONS AND ...

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