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a vocab list The*

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atoms bond Atoms
bond to gain stability
within their electrons-
In covalent
substances, atoms
share electrons to
achieve this-Atoms
are still trying to
follow the octet rule o*

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Apart from hydrogen which follows the duet rule (wants 2 valence electrons) What is a covalent bond?

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sharing of electrons
occurs the attachment
between atoms is
called answercovalent
bond questionin a
covelant bond, the
dissociation energy is*

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page 244 Draw the

Lewis structure for

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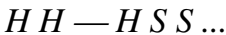
each molecule. 1. PH



respectively, for

single, double, and

triple P — — 2. H 2 S H



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bond is a shared electron pair that connects the atoms in molecular (covalent) compounds • when two atoms get close enough, their electron clouds overlap, allowing the nucleus to attract the other atom's electrons which results in a net attraction between the

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*two atoms • if atoms
are too close together,
they will repulse each*

other 8.2 Single
Covalent Bonds and
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*shared electron pairs
occupy same shell as
...*

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*shown in Figure 8.4,
each covalently
bonded atom equally
attracts the pair of
shared electrons.*

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*two atoms or ions to
be strongly attached
to each other o Ionic*

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Polyatomic ion A tightly bound group of atoms that has a positive or negative charge and behaves as a unit.

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within their electrons-

In covalent substances, atoms share electrons to achieve this-Atoms are still trying to follow the octet rule o Apart from hydrogen which follows the duet rule (wants 2 valence electrons) What is a

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Bond pages 240–247

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page 244 Draw the

Lewis structure for

each molecule. 1. PH

3 H H H H—H H P

respectively, for

single, double, and

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triple P — — 2. H 2 S H
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compounds • when

two atoms get close

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