

Chapter 18 Review Chemical Equilibrium Answers Section 1

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Chemistry: Chapter 18- Chemical Equilibrium. Equilibrium. Value of K. $K_e < 1$. $K_e > 1$. when the rates of opposing actions are the same. represents the reaction progression... -all K_e 's are dependent on.... reactants of the reaction are favored. products of the reaction are favored.

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1. In a bottle of unopened cola, the CO_2 gas dissolved in the liquid is in equilibrium with the CO_2 gas above the liquid. The dissolved gas reacts with water molecules in the cola to form carbonic acid, which also dissociates into carbon dioxide and water. Which chemical equation(s) best describe this equilibrium system? a. $CO_2(g) ? CO_2(l)$ b.

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