

## *Chapter 11 Review Gases Section 2*

~~Gas Law Problems Combined \u0026amp; Ideal Density, Molar Mass, Mole Fraction, Partial Pressure, Effusion Chapter 11 Liquids and Intermolecular Forces Chapter 10 Gases: Part 1 of 12~~

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Chapter 10 - Gases: Part 2 of 1  
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Which gas equation do I use?  
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SHORT ANSWER Answer the following questions in the space provided. b Pressure — orce For a constant force, when the surface area is tripled the surface area pressure is (a) doubled. as much. (c ripled. 7-0 (d) unchanged. Rank the following pressures in increasing

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order. (c) 76 torr (a) 50 kPa O, OO\ctbv-x

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If a gas and a liquid are the same temperature and pressure, diffusion occurs much faster in the gas because. A. there are more elastic collisions between the particles in a gas. B. gases are more compressible. C. the particles move faster in a gas and there is a greater distance between them.

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Section 11.4 Dalton's Law of Partial Pressures Goals To describe the properties of mixtures of gases. To describe

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calculations that deal with mixtures of gases. In the real world, gases are usually mixtures. This section describes how mixing gases affects the properties of the resulting mixture.

### Chapter 11 - Gases

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ANSWER Answer the following questions in the space provided.

1. b Pressure  $\propto \frac{1}{\text{area}}$ . For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged.

2. d, c, a, b Rank the following pressures in increasing order. (a) 50 kPa (c) 76 torr (b) 2 atm (d) 100 N/m<sup>2</sup>

3.

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SECTION 11.1 Key Terms pressure newton barometer millimeters of mercury

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ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: increase a. temperature increases, volume stays the same decrease b. volume increases, temperature stays the same

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following questions in the space provided. 1. \_\_\_\_\_

Pressure = . For a constant force, when the surface area  
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(c) tripled. (d) unchanged. 2. \_\_\_\_\_ Rank the following  
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in increasing order. (a) 50 kPa (c) 76 torr (b) 2 atm (d)

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ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: increase a. temperature increases, volume stays the same decrease b. volume increases, temperature stays the same

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