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A representative particle can be an atom, an ion, or a molecule, depending upon the way a substance commonly exists.

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the SI unit representing  $6.02 \times 10^{23}$  representative particles of a substance: Avogadro's number: 6.02

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$\times 10^{23}$  particles: standard temperature and pressure (00 C, 1 atm) the temperature and pressure at which one mole of gas occupies a volume of 22.4 L: molar volume: volume of a gas that contains one mole of the gas, is 22.4 L at STP:

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substance. the temperature and pressure at which one mole of gas occupies a volume of 22.4 L. equal volumes of gases at the same temperature and pressure contain equal numbers of particles.



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Chapter 10 Chemical Quantities 91

SECTION 10.1 THE MOLE: A  
MEASUREMENT OF MATTER

(pages 287–296) This section defines  
the mole and explains how the mole is

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used to measure matter. It also teaches you how to calculate the mass of a mole of any substance.

~~SECTION 10.1 THE MOLE: A  
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Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_\_ 1. What is standard temperature and pressure, or STP? a.  $0^{\circ}\text{C}$  and 101.3 kPa b.  $0^{\circ}\text{C}$  and

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1 Pa c.  $20^{\circ}\text{C}$  and 1 Pa d.  $100^{\circ}\text{C}$  and  $101.3\text{ kPa}$  \_\_\_\_\_ 2. At STP, a student measures the volume of 1.00 mole of a gas to be 22.4 L. Based on this measurement, what can the student ...

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A representative particle can be an atom, an ion, or a molecule, depending upon the way a substance commonly exists.

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