

## Atomic Timeline Answers

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~~Atomic theory Timeline~~ ~~Timeline Description: Atomic theory is the scientific theory of the nature of matter. The theory states that matter is made up of small particles called atoms. Prior to this theory, matter was thought to be able to be divided into any small quantity. The word atom is derived from the Greek atmos, meaning indivisible.~~

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~~Atomic Theory Timeline Questions and Rubric~~ ~~Please choose the correct answer to the following questions. Use the Google form to answer the questions. Answer in complete sentences. 1) Who discovered the electron?~~

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~~of atomic theory. 1. In each box, write the name of the scientist(s) associated with the statement. Choose from among the following scientists: Democritus, Rutherford, Thomson, Dalton, Bohr, Schrödinger, and Heisenberg. James Chadwick 2. On a separate sheet of paper(s), construct a timeline, and label the following: 440 BC, 1803, 1897,~~

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~~Atomic Theory Timeline~~ ~~Scientific knowledge builds on past research and experimentation. Democritus 400 B.C. Information Atomic Model Analogy Democritus, a philosopher in ancient Greece, began the search for a description of matter. He questioned whether matter could be~~

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~~History Of Atomic Theory Webquest Answer Key~~

John Dalton, a British scientist, proposed his own atomic theory. 1.) Each element is composed of extremely small particles called atoms. 2.) All atoms of a given element are identical, but the atoms of one element are different from the atoms of all other elements. 3.)

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August 12, 1887 - January 4, 1961 In 1987 J.J. Thomson discovered that the solid ball model of an atom was incorrect because there were negatively charged particles within the atom as well as positive during the Cathode Ray Tube Experiment. He gave the atom model it's electrons, calling it the Plum Pudding Atomic Model.

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1. Elements are composed of small indivisible particles called atoms. 2. Atoms of the same element are identical. Atoms of different elements are different. 3. Atoms of different elements combine together in simple proportions to create a compound. 4. In a chemical reaction, atoms are rearranged, but not changed.

~~Use the following information to fill out your foldable ...~~

During the period 1803-1807, a chemical atomic theory was developed by an English schoolteacher named John Dalton. Reasoning from a large number of observations, Dalton made four statements...

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