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Intermolecular Forces and
Properties AP Chemistry:
3.1-3.3 Intermolecular
Forces, Solids, Liquids, and
Gases Chapter 3 -
Stoichiometry and
Calculations with Formulas
and Equations: Part 1 of 5
Chapters 1 - 3 Practice Test
AP Chemistry: 3.7-3.10**

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Solubility AP Chemistry
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Molecular Theory Chapter 3
- Chemical Reactions and
Reaction Stoichiometry
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AP chemistry exam -- tips and tricks Stoichiometry Made Easy: The Magic Number Method AP Chemistry Unit 2 Review: Compound Structure and Properties (includes dot structure stuff :D) Stoichiometry Tutorial: Step by Step Video + review problems explained | Crash Chemistry Academy AP Chemistry 2020 Exam Prediction Chapter 4 Reactions in Aqueous Solution (Sections 4.1 - 4.4) Zumdahl Chemistry 7th ed. Chapter 6 (Part 1) An Intro to Chemical Reactions: Chapter 3 - Part 1 General

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College Chem Final Exam
AP Chemistry - Unit 1
Review AP Chemistry:
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Concentration Changes AP
Chem, Chapter 3, Lecture 1
AP Chemistry Unit 1
Review: Atomic Structure
and Properties!! BBrown AP
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Chapter 3 Test
AP Chemistry - Practice
Test: Chapter 3 Name _____
MULTIPLE CHOICE. Choose
the one alternative that
best completes the~~**

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statement or answers the question. 1) Silver has two naturally occurring isotopes with the following isotopic masses: 107.47 amu and 108.9047 amu . The average atomic mass of silver is 107.8682 amu .

AP Chemistry - Practice Test: Chapter 3
AP Chemistry Test (Chapter 3) (November) Multiple Choice and FIB (40%)
1) A chemistry student is filtering and drying a precipitate that formed from two solutions reacting. Which one is the most likely reason for an

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exceptionally high % yield?

A) Careless filtering and insufficient drying B)

Careless filtering and sufficient drying

AP Chemistry Test (Chapter 3)

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Name _____ MULTIPLE

CHOICE. Choose the one

alternative that best

completes the statement or answers the question.

1) Silver has two naturally occurring isotopes with the following isotopic masses:

107.47 Ar 107.47 Ar

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106.90509 108.9047 The average atomic mass of silver is 107.8682 amu.

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3 1. In 1928, 3.3 g of a new element was isolated from 660 kg of the ore molybdenite. The percent by mass of this element in the ore was: a. 5 % b. 6.6 % c. 3.3 % d. 5.0×10^{-4} % e. 2.2 %

2. A quantitative observation a. contains a number and a unit. b. does not contain a number.

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Practice Questions Multiple
Choice Identify the choice
that best completes the
statement or answers the
question. ____ 1.
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0.800 grams of an unknown hydrocarbon yields 2.613 g CO₂ and 0.778 g H₂O. What is the percent composition of the hydrocarbon? a.

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Test: Chapter 3 Name _____

MULTIPLE CHOICE. Choose

the one alternative that

best completes the

statement or answers the

question. 1)Silver has two

naturally occurring isotopes

with the following isotopic

masses: 107.47 Ar 107.47

Ar 106.90509 108.9047 The

average atomic mass of

silver is 107.8682 amu.

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A) Careless filtering and insufficient drying

B) Careless filtering and sufficient drying

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Name _____ MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

1) Silver has two naturally occurring isotopes with the following isotopic masses: 107.90509 amu and 108.9047 amu . The average atomic mass of silver is 107.8682 amu .

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1. In 1928, 3.3 g of a new element was isolated from 660 kg of the ore molybdenite. The percent by mass of this element in the ore was: a. 5 % b. 6.6 % c. 3.3 % d. 5.0×10^{-4} % e. 2.2 %

2. A quantitative observation a. contains a number and a unit. b. does not contain a number.

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